

# Development of a gas target system for highenergy resolution measurement at 0°

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A gas target system for high energy resolution measurement with using a dispersed beam at 0° was developed. The system was designed to achieve an energy resolution of 30 keV in FWHM in  $(p, p')$  experiments at 0° via a 295 MeV proton beam. A specification of the target system was modified from Ref. [1], and then a target apparatus was newly produced. An energy resolution goes worse in a measurement at finite angles because a matching condition of the Grand Raiden spectrometer can not satisfy.

A target cell has a wide window to pass a dispersed beam which results in being horizontally broadened at the target position. A size of the window was  $44 \times 14 \text{ mm}^2$  in the dispersive direction and its perpendicular one, respectively (Fig. 1 (A)). An aramid film with a thickness of  $6 \text{ }\mu\text{m}$  was used as a window material of the cell. It was found to be able to bear a pressure of up to 3 atm. A typical pressure of a target gas filled in the cell was 1 atm. Liquid nitrogen can be used to cool a gas in order to increase its target thickness.

A temperature and a pressure of a target gas were measured to deduce a target thickness. A temperature sensor of a Pt-100 and a pressure transmitter were used. The data were recorded and monitored on a Linux PC through a GPIB connection. A cell length is also an important parameter to deduce its target thickness, however, it is hard to measure it directly because a balloon like property of an aramid film increases a total length of the cell when a gas is filled up to at 1 atm. A  $\text{CO}_2$  gas was used to deduce a precise length of the cell via a cross section of  $^{12}\text{C}$  which was well known. Here, note that an aramid film also contains an element of carbon.

Because a dry pump is used in a gas handling system, a target gas can be filled into the cell without any oil contaminations. Even if a pressure of a source gas is less than 1 atm, the pump can compress it up to at 1 atm at the least, and 4 atm at the most when there is much source gas. When a connection from the cell to the pump is reversed, a gas used in the cell can be collected and stored in a bottle through the pump. Such a recycling gas system can be useful in case that one uses an expensive gas, *e.g.*  $^{36}\text{Ar}$ .

Figure 1 (B) shows a typical spectrum of the cell windows, aramid, with an empty target gas at 0°. We note that there are two types of aramid, one is made by Tore and its compositional formula is  $\text{C}_{49}\text{O}_7\text{N}_7\text{Cl}_6\text{H}_3$ , and the other is by Asahi-kase and  $\text{C}_{14}\text{O}_2\text{N}_2\text{H}_3$ . A thin film, we used, includes chlorine. Typical results with filling a target gas,  $^{20}\text{Ne}$  and  $^{36}\text{Ar}$ , are shown in Ref. [2].

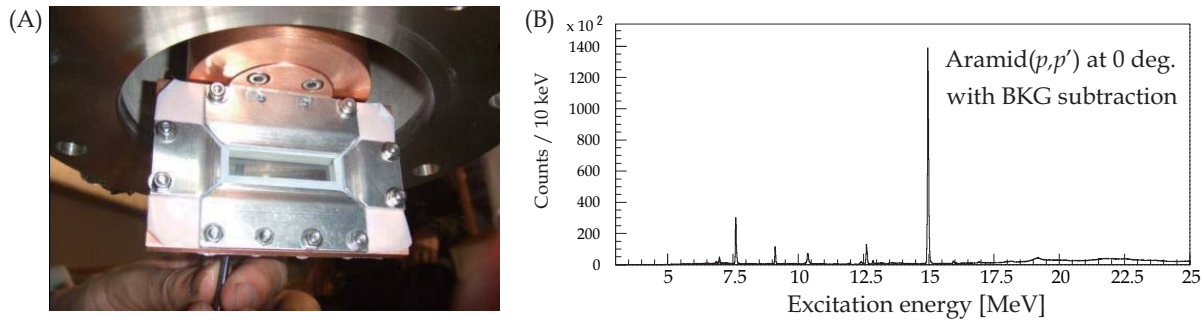


Figure 1: (A) : A photograph of the gas cell newly developed is shown. It has a wide window to pass a dispersed beam. (B) : A typical spectrum of the cell with an empty gas at 0° with background subtraction is shown. Signals due to aramid films ( $\text{C}_{49}\text{O}_7\text{N}_7\text{Cl}_6\text{H}_3$ ) are seen.

## References

- [1] T. Kawabata *et al.*, Nucl. Instr. Meth. A 459 (2001) 171.
- [2] H. Matsubara *et al.*, in this Annual Report.